415. A Synthesis of 3-Acyl- and 3-Aroyl-mesobenzanthrones.*

By F. IRVING and A. W. JOHNSON.

3-Acyl- and 3-aroyl-mesobenzanthrones have been prepared by the reaction of $\alpha\beta$ -ethynyl ketones and methyleneanthrone.

3-Benzoylmesobenzanthrone is an important dye intermediate, for on cyclisation by heating with aluminium chloride-sodium chloride in the presence of oxygen it yields 1:2:6:7-dibenzpyrene-3:8-quinone (Indanthrene Golden Yellow GK; I). It is therefore of interest that a simple method has been found for the preparation of 3-aroylmesobenzanthrones which are potential sources of dibenzpyrenequinones substituted in one of the terminal rings. The Diels-Alder reaction of methyleneanthrone and $\alpha\beta$ -ethynyl ketones proceeds readily in acetic acid solution on warming to give 3-acyl- and 3-aroyl-mesobenzanthrones in 50—70% yields, the addition being accompanied by oxidation:

$$(I.)$$

$$R = Me, Ph, o-Cl\cdot C_6H_4, p-Cl\cdot C_6H_4, m-MeO\cdot C_6H_4, p-MeO\cdot C_6H_4, \alpha-C_{10}H_7.$$

The $\alpha\beta$ -ethynyl ketones required for the Diels-Alder reactions were prepared by chromic acid oxidation of the corresponding carbinols (Bowden, Heilbron, Jones, and Weedon, J., 1946, 39; Bowden and Jones, ibid., p. 52). The condensation of methyleneanthrone with β -chloropropiophenone (which readily loses hydrogen chloride to give phenyl vinyl ketone) in nitrobenzene solution to give 3-benzoylmesobenzanthrone has been described by Allen, Bell, Bell, and van Allan (J. Amer. Chem. Soc., 1940, 62, 656) although extensions of this method would be restricted by the limited availability of the starting materials.

EXPERIMENTAL.

Preparation of Intermediates:—p-Methoxyphenylethynylcarbinol. Purified acetylene was rapidly passed into liquid ammonia (1,000 c.c.) with stirring and cooling, and sodium (23 g.) added in small pieces in such a manner that the blue colour never persisted for more than a few moments. A solution of p-methoxybenzaldehyde (redistilled: 136 g.) in ether (100 c.c.) was then added during $1\frac{1}{2}$ hours and the mixture stirred and cooled for a further 3 hours with continued introduction of acetylene. The ammonia was allowed to evaporate overnight, and dilute sulphuric acid (ice) was added to the residue. The product was extracted with ether (3 × 300 c.c.) and the ethereal extracts washed with sodium hydrogen carbonate solution and dried. After removal of the solvent the residue was distilled to yield p-methoxyphenylethynylcarbinol (71 g.) as a pale yellow oil, b. p. 146—149°/15 mm. (Found: C, 73·7; H, 6·05. $C_{10}H_{10}O_2$ requires C, 74·05; H, 6·2%). The product readily formed a white insoluble silver derivative.

* Patent application pending.

Since the above work was carried out, Rutan and May (J. Amer. Chem. Soc., 1947, 69, 2017) have described p-methoxyphenylethynylcarbinol as a crystalline solid, m. p. 172—173°. This may be a misprint (phenylethynylcarbinol has m. p. 22°). The sample obtained in the above experiment was oxidised directly to the corresponding ketone and no special effort was made to crystallise it.

p-Methoxyphenyl ethynyl ketone. To a stirred solution of p-methoxyphenylethynylcarbinol (15·6 g.)

in acetone (25 c.c.), a solution of chromium trioxide (8.0 g.) in water (30 c.c.) and concentrated sulphuric acid (7 c.c.) was slowly added. The operation, carried out at 5—10° in nitrogen, lasted 1½ hours. After being stirred for a further 30 minutes, the mixture was diluted with water and the product extracted with ether. Evaporation of the ethereal solution gave a yellow solid which was crystallised from aqueous methanol and then petroleum (b. p. 60—80°) to give p-methoxyphenyl ethynyl ketone (9·1 g.) as pale yellow needles, m. p. 82—83° (Found: C, 74·7; H, 4·7. $C_{10}H_8O_2$ requires C, 75·0; H, 5·0%). The ketone readily formed a white insoluble silver derivative and a brick-red insoluble copper derivative.

a-Naphthyl ethynyl ketone. a-Naphthylethynylcarbinol was prepared from a-naphthaldehyde (35.6 g.; Badger, J., 1941, 536) by the method described above for p-methoxyphenylethynylcarbinol. The crude product (34 g.) was not purified but dissolved in acctone (50 c.c.) and oxidised with a solution

of chromium trioxide (18 g.) was not purned but dissolved in acctone (30 c.c.) and oxidised with a solution of chromium trioxide (18 g.) in water (56 c.c.) and concentrated sulphuric acid (15 c.c.) as described for the preparation of p-methoxyphenyl ethynyl ketone. The crude ketone (24 g.) was crystallised from petroleum (b. p. 60—80°) to yield a-naphihyl ethynyl ketone as pale yellow plates, m. p. 70—71° (Found: C, 86·25; H, 4·7. C₁₃H₈O requires C, 86·65; H, 4·45%).

Preparation of 3-Acyl- and 3-Aroyl-mesobenzanthrones.—3-Benzoylmesobenzanthrone. Methylene-anthrone (1·3 g.; Clar, Ber., 1936, 69, 1686) and phenyl ethynyl ketone (1 g.; Bowden, Heilbron, Jones, and Weedon, J., 1946, 39) were dissolved in acctic acid (6 c.c.) and heated under reflux for \(\frac{3}{4}\) hour. The colution at first became dark brown and then lightened until it was finally golden yellow. solution at first became dark brown and then lightened until it was finally golden yellow. 3-Benzoyl-mesobenzanthrone (1·2 g.), m. p. 186—188°, was deposited on cooling and was crystallised from acetic acid, being obtained as bright yellow prisms, m. p. 193-194° (Allen, Bell, Bell, and van Allan, loc. cit., give 192°).

3-Acetylmesobenzanthrone. Prepared similarly from methyleneanthrone (5 g.) and methyl ethynyl ketone (1.75 g.; Bowden, Heilbron, Jones, and Weedon, loc. cit.) in acetic acid (25 c.c.), 3-acetylmesobenzanthrone (3.0 g.) formed bright yellow platelets, m. p. 173—175°, from aqueous acetic acid (I.G. Farben., B.P. 440,263, give m. p. 174—175°).

3-p-Methoxybenzoylmesobenzanthrone. Prepared similarly from methyleneanthrone (3.9 g.) and p-methoxyphenyl ethynyl ketone (3.07 g.) in acetic acid (24 c.c.), 3-p-methoxybenzoylmesobenzanthrone (2.9 g.) formed yellow prisms, m. p. 194—195°, from acetic acid (Found: C, 82.0; H, 4.6. C₂₅H₁₆O₃ requires C, 82.4; H, 4.4%).

3-m-Methoxybenzoylmesobenzanthrone. Prepared similarly from methyleneanthrone (11-1 g.) and m-methoxyphenyl ethynyl ketone (8·85 g.; Johnson and Melhuish, J., 1947, 346) in acetic acid (70 c.c.).
3-m-Methoxybenzoylmesobenzanthrone after crystallisation from acetic acid formed bright yellow prisms (8·85 g.), m. p. 192—193° (Found: C, 82·8; H, 4·7%).
3-o-Chlorobenzoylmesobenzanthrone. Methyleneanthrone (62·5 g.) and o-chlorophenyl ethynyl ketone

(50 g., Johnson, f., 1947, 1626) in acetic acid (750 c.c.) were heated on the steam-bath for 1½ hours. The solution at first became dark and then lightened as the dehydrogenation occurred. 3-o-Chlorobenzoylmesobenzanthrone (49 g.) separated from the solution on cooling and a further quantity (13 g.) was obtained by cautious dilution of the mother liquors. Crystallisation from acetic acid gave the product as yellow prisms, m. p. 206—207° (Found: C, 78·25; H, 3·75. C₂₄H₁₃O₂Cl requires C, 78·2;

H, 3.5%).

3-a-Naphthoylmesobenzanthrone. Prepared similarly from methyleneanthrone (9·1 g.) and α-naphthyl ethynyl ketone (8 g.) in acetic acid (60 c.c.), the product (7·6 g.) was sparaged and crystallised from 1.254-255° (I.G. chlorobenzene to yield 3-a-naphthoylmesobenzanthrone as dark yellow prisms, m. p. 254—255° (I.G. Farben., D.R.-P. 446,187, give m. p. 244—245°) (Found: C, 87.0; H, 4.15. Calc. for $C_{28}H_{16}O_2$: C, 87.5; H, 4.2%). The product dissolved in concentrated sulphuric acid to a cherry-red solution.

IMPERIAL CHEMICAL INDUSTRIES LTD., RESEARCH LABORATORIES HEXAGON HOUSE, BLACKLEY, MANCHESTER, 9. [Received, February 16th, 1948.]